

SIR C.R.REDDY COLLEGE FOR WOMEN, ELURU
CURRICULUM LECTURER WISE 2022 - 2023

ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: R. Lakshmi Tulasi					CLASS: IMPET ZFC		Semester: I		Paper: Inorganic & Physical chemistry				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
October	3 rd	02	Syllabus dictation, Unit-I:- Introduction of P-block elements.	Types of chemical bonds,	-	-	-	-	-	-	-	-	-
	4 th	05	Group-13: Preparation and structure of diborane, Borazine	metals and Non metals.	-	-	-	-	-	-	-	-	-
			Preparation, classification and uses of silicones.	-	-	-	-	-	-	-	-	-	-
			Preparation and structure of phosphonic acids.	-	-	-	-	-	-	-	-	-	-
			($PNCl_2$) _n where n = 3, 4	-	-	-	-	-	-	-	-	-	-
November	1 st	05	oxides and acids of sulphur (structures only)	-	-	-	-	-	-	-	-	-	-

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ANNUAL CURRICULAR					PLAN (Year)							
NAME OF THE LECTURER: <u>R. Lakshmi Tulasi</u>					CLASS: <u>IAPE + ZFc</u>			Semester: <u>I</u>		Paper: <u>In organic and physical chemistry</u>		
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY			
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date
			Pseudohalogen, structures of Inter halogen compounds.	-	-	-	-	-	-	-	-	-
			unit-II: characteristics of d-block elements	-	Remedial	01	Yes	-	-	-	-	-
			with special reference to e.c, variable valence.	-	-	-	-	-	-	-	-	-
November	2 nd	04	Magnetic Properties, catalytic Properties	-	-	-	-	-	-	-	-	-
			& ability to form complexes, stability of various OS.	-	-	-	-	-	-	-	-	-
	3 rd	05	Chemistry of f-block elements: chemistry of lanthanides.	-	-	-	-	-	-	-	-	-

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ANNUAL CURRICULAR					PLAN (Year)							
NAMR OF THE LECTURER R. Lakshmi Tulas					CLASS : IMPC + ZFc		Semester : I		Paper : Inorganic and Physical Chemistry			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY			
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date
			electronic structure, O.S, Lanthanide contraction,	-	Will be shown ppt by Lanthanide contraction	01	Yes	-		-	-	-
			consequences of Lanthanide contraction,	-	-	-	-	-		-	-	-
			magnetic properties chemistry of Actinides: E.C, O.S	-	-	-	-	-		-	-	-
November	4th	06	Actinide contraction comparison of	-	Remedial	01	Yes	-		-	-	-
			Lanthanides & Actinides.	-	-	-	-	-		-	-	-
			Theories of Bonding in metals: V.B.T & free e theory,	-	-	-	-	-		-	-	-

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ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: <u>R. Lakshmi Tulas</u>					CLASS: <u>TMPC + ZFC</u>			Semester: <u>T</u>		Paper: <u>Inorganic and Physical Chemistry</u>			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
			Explanation of the thermal & electrical	-	-	-	-	-	-	-	-	-	-
			conductivity of metals based on these theories.	-	01	Yes	-	-	-	-	-	-	-
			Band theory-formation of bands, exp.	-	-	-	-	-	-	-	-	-	-
			Explanation of conductors, semiconductors & insulators.	-	-	-	-	-	-	-	-	-	-
Decem-ber	st	05	Unit-III: symmetry in crystals. The law of symmetry.	-	will be shown PPT by symmetry	01	Yes	-	-	-	-	-	-
			Law of constancy of interfacial angles The law of rationality	-	-	-	-	-	-	-	-	-	-

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ANNUAL CURRICULAR					PLAN (Year)				Inorganic and Physical Chemistry				
NAME OF THE LECTURER: <u>R. Lakshmi Tulasi</u>					CLASS: <u>TMPC + ZFC</u>		Semester: <u>I</u>		Paper: <u>Inorganic and Physical Chemistry</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
			of indices, Miller indices. Definition of lattice Point,	-	-	-	-	-	-	-	-	-	-
			Space lattice, unit cell. Bravais lattices & crystal systems.	-	-	-	-	-	-	-	-	-	-
	2 nd	02	X-ray diffraction & crystal structure Bragg's law.	-	Remedial	01	Yes	-	-	-	-	-	-
			Powder methods. Defects in crystals stoichiometric &	-	-	-	-	-	-	-	-	-	-
			Non-stoichiometric defects.	-	-	-	-	-	-	-	-	-	-
	3 rd	05	1. Gaseous state:- vanderwaal's eq ⁿ of state.	-	-	-	-	-	-	-	-	-	-

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NAME OF THE LECTURER: R. Lakshmi Tulasī					CLASS: IMPC + ZFC				Semester: I				Paper: Inorganic and Physical chemistry			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY							
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date				
			Andrew's isotherms of CO_2 , continuity of state.	-	debate	01	Yes	-	-	-	-	-	-			
			critical phenomena, Relationship b/n critical constants and	-	-	-	-	-	-	-	-	-	-			
			Vanderwaal's constant law of corresponding states.	-	-	-	-	-	-	-	-	-	-			
Dec	4th	06	Joule-Thomson effect. Inversion temperature.	-	will be shown PPT by Joule-Thomson effect.	01	Yes	-	-	-	-	-	-			
			liquid state:- Liquid crystals, mesomorphic state.	-	-	-	-	-	-	-	-	-	-			
			Differences b/n liquid crystal & solid/liquid.	-	-	-	-	-	-	-	-	-	-			

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NAME OF THE LECTURER: <u>R. Lakshmi Tulasi</u>					CLASS: <u>IMPc + zFc</u>		Semester: <u>I</u>		Paper: <u>Inorganic and Physical chemistry</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
			classification of liquid crystals into smectic & Nematic.	-	-	-	-	-	-	-	-	-	-
			Appn of Liquid crystals as LED devices.	-	-	-	-	-	-	-	-	-	-
			<u>Unit - I</u> : Solutions:- Azeotropes-HCl, H ₂ O and ethanol-water system.	Definitions of concentration terms.	-	-	-	-	-	-	-	-	-
Dec	5 th	01	and ethanol-water system.	Types of binary solns. Types of liq-liq solns.	-	-	-	-	-	-	-	-	-
Jan	1 st	05	Partially miscible liquids-Phenol H ₂ O system.	-	Remedial	01	Yes	-	-	-	-	-	-
			critical solution temperature (CST) effect of impurity	-	-	-	-	-	-	-	-	-	-

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NAME OF THE LECTURER <u>R. Lakshmi Tulasi</u>					CLASS : <u>IMPC + ZFC</u>		Semester : <u>I</u>		Paper : <u>Inorganic and Physical Chemistry</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
			on consolute temperature. Immiscible liquids & steam	-	-	-	-	-	-	-	-	-	-
			distillation. Nernst distribution law.	-	Group discussion	01	yes	-	-	-	-	-	-
			calculation of the Partition coefficient.	-	-	-	-	-	-	-	-	-	-
			Applications of distribution law.	-	-	-	-	-	-	-	-	-	-
	2nd	02	<u>Ionic equilibrium</u> :- Ionic Product, com-	-	-	-	-	-	-	-	-	-	-
			mon ion effect & its applications.	-	-	-	-	-	-	-	-	-	-

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NAME OF THE LECTURER: <u>R. Lakshmi Tulas</u>					CLASS: <u>INPC+2FC</u>		Semester: <u>I</u>		Paper: <u>Inorganic and Physical Chemistry</u>				
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	3 rd	03	solubility & solubility product. calculations based on solubility product.	-	-	-	-	-	-	-	-	-	-
			Dilute solutions + colligative properties	-	01	Yes	-	-	-	-	-	-	-
			RLVP, Raoult's law. osmotic pressure,	-	-	-	-	-	-	-	-	-	-
	4 th	06	elevation in B.P & depression in F.P	-	01	Yes	-	-	-	-	-	-	-
			experimental methods for the determination of molar mass of a non-volatile solute using	-	-	-	-	-	-	-	-	-	-

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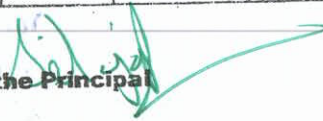
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MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
			osmotic pressure. elevation in B.P & depression in F.P	-	will be shown PPT by elevati- on in B.P	01	Yes	-	-	-	-	-	-
Feb	1st	05	Abnormal colligative properties.	-	-	-	-	-	-	-	-	-	-
			vant Hoff factor	-	-	-	-	-	-	-	-	-	-
	2nd	05	Revision for II Mid exams.	-	-	-	-	-	-	-	-	-	-
	3rd	05	II Mid exams	-	-	-	-	-	-	-	-	-	-
	4th	05	Revision	-	-	-	-	-	-	-	-	-	-

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ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: <u>K. SUJATHA</u>					CLASS: <u>II B.Sc</u>		Semester: <u>III</u>		Paper: <u>III</u> <u>Organic chemistry & Spectroscopy</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
Oct	4 th	2	<u>UNIT-1: Syllabus dictation, Introduction</u>	-	-	-	-	-	-	-	-	-	-
Nov	1 st	5	<u>Methods of preparation and properties, nucleophilic substitution reactions with stereochemical aspects and effect of solvent, nucleophilic substitution vs elimination, Williamson's synthesis, Aryl halides - preparation & properties, nucleophilic aromatic substitution. S_N, A₂.</u>	-	-	-	-	-	-	-	-	-	-

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SIR C.R.REDDY COLLEGE FOR WOMEN, ELURU
CURRICULUM LECTURER WISE 20²²-2023

ANNUAL CURRICULAR					PLAN (Year) <i>Organic chemistry & spectroscopy</i>								
NAME OF THE LECTURER <i>K. SUTATHA</i>					CLASS : <i>II B.Sc</i>			Semester : <i>II</i>		Paper : <i>III</i>			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
<i>NOV</i>	<i>2</i>	<i>4h</i>	<i>Benzyne mechanism Relative reactivity of alkyl, allyl, benzyl</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>Vinyl and aryl halides towards nucleophilic substitution reactions</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>Alcohols & phenols: Alcohols preparation & properties and relative reactivity of 1^o, 2^o, 3^o alcohols, Bouveault-Blanc</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>reduction, oxidation of diols by per-iodic acid and</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>lead tetra acetate, pinacol-pinacolone rearrangement</i>	<i>Remedial class</i>	<i>-</i>	<i>01</i>	<i>yes</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>

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CURRICULUM LECTURER WISE 20~~22~~²³ - 2023

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NAMR OF THE LECTURER <u>K. SUJATHA</u>					CLASS : <u>II BSC</u>		Semester : <u>III</u>		Paper : <u>III</u> <u>Organic chemistry & spectroscopy</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
<u>NOV</u>	<u>3</u>	<u>5</u>	<u>phenols: preparation and properties, acidity and factors effecting it, ring substitution reactions Reimer-Tiemann and kolbe-schmidt reactions, Fries and claisen rearrangement with mechanisms</u>	-	-	-	-	-	-	-	-	-	-
			<u>UNIT-II: Carbonyl compounds: structure reactivity, preparation and properties nucleophilic addition and nucleophilic addition elimination reactions with ammonia derivative</u>	-	-	-	-	-	-	-	-	-	-
				-	<u>Remedial class</u>	<u>01</u>	<u>yes</u>	-	-	-	-	-	-

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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year) Organic chemistry & spectroscopy								
NAME OF THE LECTURER K. SUJATHA					CLASS : II B.Sc			Semester : III		Paper : III			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
NOV	4	5	Mech of aldol & benzoin condensation	-	-	-	-	-	-	-	-	-	-
			elaisen schmidt, perkin, Cannizzaro and wittig reaction	-	-	-	-	-	-	-	-	-	-
			Beckmann and Haloform reaction, and Bayer-villiger oxidation, α -	-	-	-	-	-	-	-	-	-	-
			Substitution reactions	-	-	-	-	-	-	-	-	-	-
			clemensen reduction, wolf-kishner, with LiAlH ₄ NaBH ₄ reductions addition reactions of d, B-unsaturated carbonyl compounds,	-	-	-	-	-	-	-	-	-	-
			Michael addition	-	-	-	-	-	-	-	-	-	-
				-	-	-	-	-	-	-	-	-	-

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NAME OF THE LECTURER: <u>K. SUTATHA</u>					CLASS: <u>II B.Sc</u>		Semester: <u>III</u>		Paper: <u>III</u> <i>Organic chemistry & Spectroscopy</i>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
Dec	1	5h	Active methylene compounds: keto enol tautomerism	-	-	-	-	-	-	-	-	-	-
			preparation and synthetic applications of diethyl malonate and ethyl aceto acrylate	-	-	-	-	Quiz	01	Yes	-	-	-
			malonate and ethyl aceto acrylate	-	-	-	-	-	-	-	-	-	-
			UNIT-III: Carboxylic acids and their derivatives.	-	-	-	-	-	-	-	-	-	-
			General methods of preparation, physical properties and reactions of mono carboxylic acids, effect	-	-	-	-	-	-	-	-	-	-
				-	Revision	01	Yes	-	-	-	-	-	-

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CURRICULUM LECTURER WISE 20²²-2023

ANNUAL CURRICULAR					PLAN (Year) <i>organic chemistry & spectroscopy</i>								
NAME OF THE LECTURER <i>K. SUJATHA</i>					CLASS : <i>II B-SC</i>		Semester : <i>III</i>		Paper : <i>III</i>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
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			<i>of substituent on acidic strength</i>	—	—	—	—	—	—	—	—	—	—
			<i>Typical reactions of dicarboxylic acids</i>	—	—	—	—	—	—	—	—	—	—
<i>Dec</i>	<i>2</i>	<i>4h</i>	<i>Hydroxy acids & unsaturated acids prep & reactions of</i>	—	—	—	—	—	—	<i>Group discussion</i>	<i>01</i>	<i>Yes</i>	—
			<i>acid chlorides, anhydrides, esters and amides</i>	—	—	—	—	—	—	—	—	—	—
			<i>Comparative study of nucleophilic substitution at acyl</i>	—	—	—	—	—	—	—	—	—	—
			<i>group - mechanism of acidic and alkaline hydrolysis of esters</i>	—	—	—	—	—	—	—	—	—	—

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MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
<i>Dec</i>	<i>3</i>	<i>5h</i>	<i>claisen condensation, Reformatsky reaction and</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>Curtius rearrangement reactions involving H, OH, &c</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>COOH groups, salt formation, anhydride formation, acid chloride formation, amide formation and esterification, Degradation of carboxylic acids by Huns-Dieckmann reaction, Schmidt reaction</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>
			<i>Arndt-Eistler synthesis HVZ reaction</i>	<i>-</i>	<i>Remedial class</i>	<i>01</i>	<i>yes</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>-</i>

K. Sujatha
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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year) organic chemistry & spectroscopy									
NAME OF THE LECTURER K-SOJATHA					CLASS : <u>II B.Sc</u>				Semester : <u>III</u>		Paper : <u>III</u>		CO-CURRICULAR ACTIVITY	
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date		
Dec	4	4	UNIT- IV : Rotation spectroscopy selection rules, intensities of spectral lines, determination of bond lengths of diatomic and linear triatomic molecules. Isotopic substitution.	-	-	-	-	-	-	-	-	-		
				-	-	-	-	-	-	-	-	-		
				-	-	-	-	-	-	-	-	-		
				-	-	-	-	-	-	-	-	-		
Jan	1	5	Vibrational spectroscopy classical equation of vibration, computation of force constant harmonic and	-	Remedial class	01	Yes	-	-	-	-	-		
				-	-	-	-	-	-	-	-	-		

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B. D.
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CURRICULUM LECTURER WISE 2022-2023

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ANNUAL CURRICULAR					PLAN (Year) <i>organic chemistry & spectroscopy</i>									
NAME OF THE LECTURER <i>K. SUJATHA</i>					CLASS : <i>II B-sc</i>				Semester : <i>IV</i>		Paper : <i>III</i>			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY					
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date		
			<i>and anharmonic oscillator</i>	-	-	-	-	-	-	-	-	-	-	
<i>Jan</i>	<i>2</i>	<i>2</i>	<i>Morse potential curve, vibrational degree of freedom</i>	-	-	-	-	-	-	-	-	-	-	
			<i>for polyatomic molecules, modes of vibration</i>	-	-	-	-	-	-	-	-	-	-	
	<i>3</i>	<i>4</i>	<i>selection rules for vibrational transitions</i>	-	-	-	-	-	-	-	-	-	-	
			<i>Fundamental frequencies, overtones and hot bands</i>	-	-	-	-	-	-	-	-	-	-	
			<i>Electronic Spectroscopy: energy levels of molecular orbitals</i>	-	-	-	-	-	-	-	-	-	-	
			<i>Selection rules for</i>	-	-	-	-	-	-	-	-	-	-	

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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year) <i>organic chemistry & spectroscopy</i>							
NAME OF THE LECTURER <i>K. SUJATHA</i>					CLASS : <i>II B-SC</i>			Semester : <i>III</i>		Paper : <i>III</i>		
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY			
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date
			<i>electronic spectra types of electronic spectra</i>	-	-	-	-	-	-	-	-	-
<i>Jan</i>	<i>4</i>	<i>4</i>	<i>effect of conjugation chromophore, batho chromic and</i>	-	-	-	-	-	-	-	-	-
			<i>hypso chromic shifts Beer-Lambert's law and E_t</i>	-	-	-	-	-	-	-	-	-
<i>Feb</i>	<i>1</i>	<i>5h</i>	<i>limitations NMR spectroscopy; principles of nuclear magnetic resonance equivalent and non equivalent</i>	-	<i>Remedial class</i>	<i>01</i>	<i>yes</i>	-	-	-	-	-
			<i>proton, position of signals, chemical shift, NMR.</i>	-	-	-	-	-	-	-	-	-

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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year) <i>Organic Chemistry & Spectroscopy</i>								
NAME OF THE LECTURER <i>K. Sujatha</i>					CLASS : <i>ITB-SC</i>				Semester : <i>III</i>		Paper : <i>III</i>		
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
			<i>splitting of signals - spin-spin coupling, coupling constants.</i>	-	-	-	-	-	-	-	-	-	-
<i>Feb</i>	<i>2</i>	<i>5</i>	<i>Applications of NMR with ethyl bromide, ethanol</i>	-	-	-	-	-	-	-	-	-	-
			<i>acetaldehyde, 1,1,2-tribromo ethane, ethyl</i>	-	-	-	-	-	-	-	-	-	-
			<i>acetate toluene and acetophenone.</i>	-	-	-	-	-	-	-	-	-	-
<i>Feb</i>	<i>3</i>	<i>3</i>	<i>Application of visible, ultraviolet & IR spectroscopy:-</i>	-	-	-	-	-	-	-	-	-	-
			<i>Application of electronic spectroscopy and</i>	-	-	-	-	-	-	-	-	-	-

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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year) organic chemistry & spectroscopy							
NAMR OF THE LECTURER K. Sujatha					CLASS : III B.Sc			Semester : III		Paper : III		
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY			
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date
Feb	4	5	wood ward rules for calculating max of conjugated dienes and di β -unsaturated compounds.	-	-	-	-	-	-	-	-	-
			IR and types of molecular vibrations, functional group and fingerprint region. IR spectra of alkanes	-	-	-	-	-	-	-	-	-
			alkenes and simple alcohols and different functional groups	-	-	-	-	-	-	-	-	-
			revision	-	01	Yes	-	-	-	-	-	-
			alkenes and simple alcohols and different functional groups	-	-	-	-	-	-	-	-	-

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B. A.
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M. J.
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SIR C.R.REDDY COLLEGE FOR WOMEN, ELURU
CURRICULUM LECTURER WISE 2022 - 2023

ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: <u>V. RAJA RAJESWARI</u>					CLASS: <u>III BSc</u>		Semester: <u>V</u>		Paper: <u>7(B)</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
Nov	4	4	Chromatography - Principle, Classification,	Definitions of S.P, M.P and chromatography.	-	-	-	-	-	-	-	-	-
	5	4	nature of adsorbents, R _f value factors affecting R _f values.	Identification of unknown compound with R _f value.	-	-	-	-	-	-	-	-	-
Dec	1	3	TLC - Principle, Experimental procedure, adsorbents	Introduction of TLC	Explanation by showing TLC plates in lab	-	-	-	-	-	-	-	-
	2	4	solvents applications and advantages. Paper chromatography - Principle	-	-	-	-	-	-	-	-	-	-
	3	4	Experimental procedure, choice of paper and solvents	-	Showing the whatmann filter papers in lab	-	-	-	-	-	-	-	-
	4	3	Development - ascending, descending, radial and two dimensional chromatogram	-	-	-	-	-	-	-	-	-	-

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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: <u>V. RAJA RAJESWARI</u>					CLASS: <u>TU BSC</u>		Semester: <u>V</u>		Paper: <u>7(B)</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
	5	4	Applications of Paper Chromatography	-	-	-	-	-	-	-	-	-	-
Jan	1	4	Column chromatography - Principle classification	Introduction of Column	Explanation by showing the column in lab	-	-	-	-	-	-	-	-
	2	4	Experimental procedure stationary and mobile phase	Types of columns	-	-	-	-	-	-	-	-	-
	3	3	development of the chromatography applications.	-	-	-	-	-	-	-	-	-	-
	4	2	HPLC - Basic Principles, Instrumentation	-	showing the HPLC instrument in pharmacy college	-	-	-	-	-	-	-	-
Feb	1	4	- block diagram & applications.	-	-	-	-	-	Quiz	1	yes	-	-

V.R. Raja
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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year)							
NAME OF THE LECTURER: <u>V-RAJA RATESWARI</u>					CLASS: <u>TU BEE</u>		Semester: <u>IV</u>		Paper: <u>7(B)</u>			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY			
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date
Feb	2	3	<u>Spectrophotometry</u> - Principle Instrumentation - single beam and double beam spectrometry	Definition of absorbance, Transmittance	-	-	-	-	Seminar	1	Yes	-
	3	3	Beer-Lambert's Law - Derivation and derivations from Beer-Lambert Law, application of Beer-Lambert Law - Det. of Fe ²⁺ , Mn ²⁺ and Pb ²⁺	Molar absorptivity wavelength frequency Determination of Fe ³⁺	-	-	-	-	Group Discussion	1	Yes	-
	4	4+1	Atomic spectroscopy - Types, atomizer		-	-	-	-	Debate	1	Yes	-
Mar	1	4	atomic absorption and emission		-	-	-	-				
	2	3	Applications of AAS.		-	-	-	-				
	3	2			-	-	-	-				

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CURRICULUM LECTURER WISE 2022 - 2023

ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: <u>V. RAJA RAJESWARI</u>					CLASS: <u>III BSc</u>		Semester: <u>V</u>		Paper: <u>7(B) - Analytical methods in chemistry - 2</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO- CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
Nov	4	4	Chromatography - Principle, Classification	Definitions of S.P, M.P and chromatography.	-	-	-	-	-	-	-	-	-
	5	4	Nature of adsorbents, elements, R _f value factors affecting R _f values.	Identification of unknown compound with R _f value.	-	-	-	-	-	-	-	-	-
Dec	1	3	TLC - Principle, Experimental procedure, adsorbents	Introduction of TLC	Explanation by showing TLC plates in lab	-	-	-	-	-	-	-	-
	2	4	solvents applications and advantages. Paper chromatography - Principle	-	-	-	-	-	-	-	-	-	-
	3	4	Experimental procedure, choice of paper and solvents	-	Showing the Whatmann filter papers in lab	-	-	-	-	-	-	-	-
	4	3	Development - ascending, descending, radial and two dimensional chromatogram	-	-	-	-	-	-	-	-	-	-

V.R.Raja
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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year)								
NAME OF THE LECTURER: <u>V-RAJAPAJESWARI</u>					CLASS: <u>TU BSC</u>		Semester: <u>V</u>		Paper: <u>7(B) - Analytical methods in Chemistry-2</u>				
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY				
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	
	5	4	Applications of Paper Chromatography	-	-	-	-	-	-	-	-	-	-
Jan	1	4	Column chromatography - Principles, classification	Introduction of Column	Explanation by showing the column in lab	-	-	-	-	-	-	-	-
	2	4	Experimental Procedure stationary and mobile phase	Types of columns	-	-	-	-	-	-	-	-	-
	3	3	development of the chromatography applications.	-	-	-	-	-	-	-	-	-	-
	4	2	HPLC - Basic Principles, Instrumentation	-	Showing the HPLC instrument in pharmacy college	-	-	-	-	-	-	-	-
Feb	1	4	block diagram & applications.	-	-	-	-	-	Quiz	1	yes	-	-

V-R. Raja
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CURRICULUM LECTURER WISE 2022-2023

ANNUAL CURRICULAR					PLAN (Year)							
NAME OF THE LECTURER: <u>V-RAJA RAMESHWARI</u>					CLASS: <u>TU BSc</u>		Semester: <u>IV</u>		Paper: <u>7(B)-in chemistry-2</u> <i>analytical methods</i>			
MONTH	WEEK	HOURS AVAILABLE	SYLLABUS/ TOPIC	Additional Input/Value Addition Provided/ Taught	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITY			
					Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date	Activity Conducted	Hours Allotted	Whether Conducted	If not Alternate Date
Feb	2	3	<u>Spectrophotometry</u> - Principle Instrumentation - single beam and double beam spectrometry	Definition of absorbance Transmittance	-	-	-	-	Seminar	1	Yes	-
	3	3	Beer-Lambert's Law - Derivation and deviations from Beer-Lambert Law, application of Beer-Lambert Law - Det. of Fe^{+2} and Pb^{+2}	Molar absorptivity wavelength frequency Determination of Fe^{+3}	-	-	-	-	Group Discussion	1	Yes	-
	4	4+1	<u>Atomic spectroscopy</u> - Types, atomizer		-	-	-	-	Debate	1	Yes	-
Mar	1	4	atomic absorption and emission		-	-	-	-				
	2	3	Applications of AAS.		-	-	-	-				
	3	2			-	-	-	-				

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